

# **BTec National Extended Diploma in Applied Science Summer Independent Learning Y12-13**

## **Part 1 – Compulsory Content (pages 2-32)**

There are 3 sections to the compulsory content (Biology, Physics and Chemistry)

For each section.

1. Watch the videos and complete the notes – you may consider adding flashcards / condensed notes, so you can use them to test yourself (metacognition)
2. Complete the follow up application questions
3. Correct and improve the questions where available (mark scheme at the end of the document)

## **Part 2 – Highly Recommended (pages 33-42)**

There are 3 sections to the highly recommended content (Biology, Physics and Chemistry)

Solutions to highly recommended content (pages 44-49)

# Part 1 – Compulsory Content

## BIOLOGY

### Organs and Systems

Use the following links to support you with the questions

The Heart: <https://www.savemyexams.com/qcse/biology/aqa/18/revision-notes/2-organisation/2-2-organisation-the-cardiovascular--respiratory-system/2-2-2-the-heart/>

The Cardiac Cycle: <https://www.youtube.com/watch?v=3ZYh00GJZSI>

### Cardiovascular System

**Q1. Describe the structure and function of the different types of blood vessels**

#### MEMORISE FOR RETRIEVAL TEST

Vessel	Structure	Function
Arteries		
Capillaries		
Veins		

**Q2. Complete the table below on the ABO blood type system**

	Group A	Group B	Group AB	Group O
Red Blood Cell Type				
Antibodies in Plasma				
Antigens on red blood cell				

**Q3. What is the Rhesus system?**

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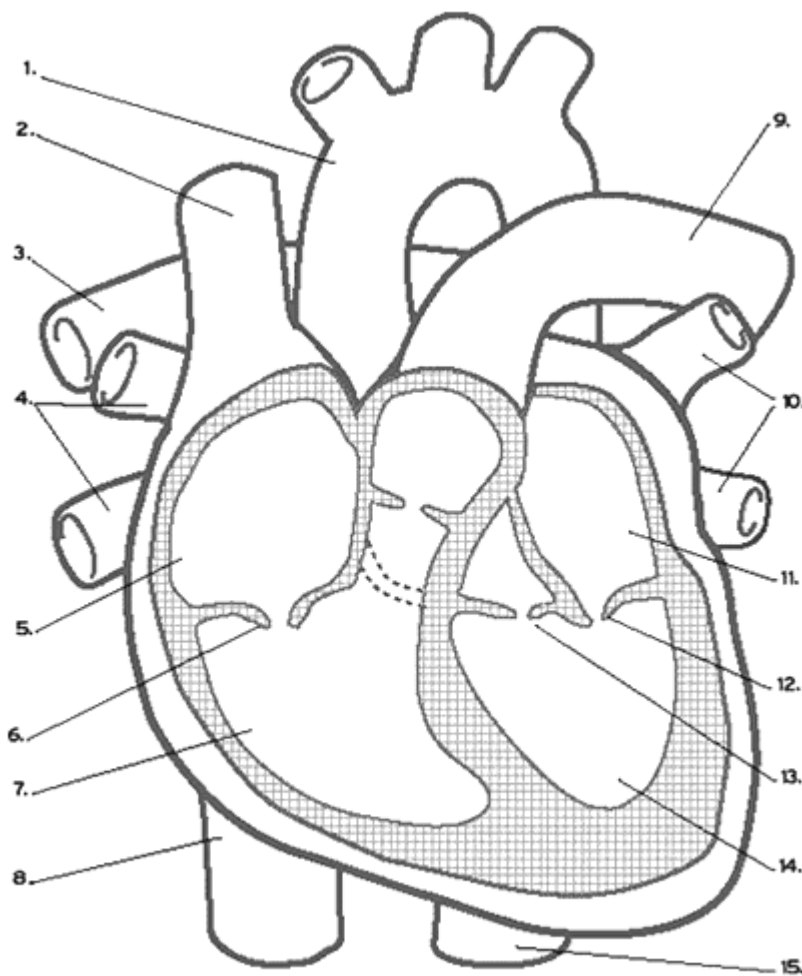
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**Q4. Label the diagram of the heart below. Shade in RED the areas where there is flow of oxygenated blood. Shade in BLUE the areas where there is flow of deoxygenated blood.**

**MEMORISE FOR RETRIEVAL TEST**



## Apply

Explain how the structure of an artery is related to its functions.

(3)

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Describe two differences between the structure of a capillary and the structure of a vein

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# Lungs and Ventilation

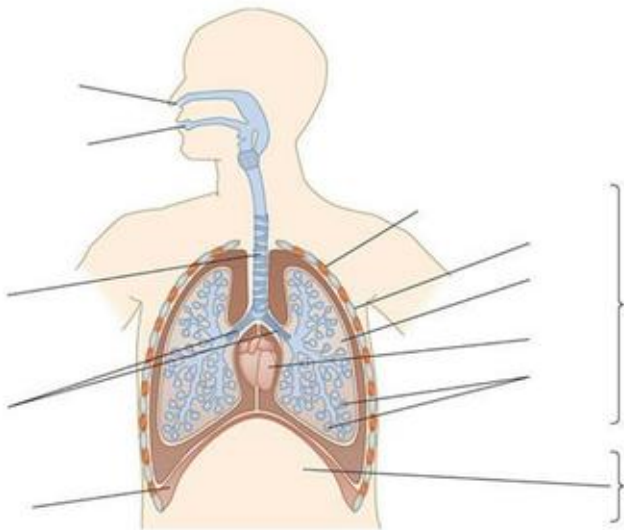
Use the following links to support you with the questions

The Lungs: <https://www.savemyexams.com/qcse/biology/aqa/18/revision-notes/2-organisation/2-2-organisation-the-cardiovascular--respiratory-system/2-2-1-the-lungs/>

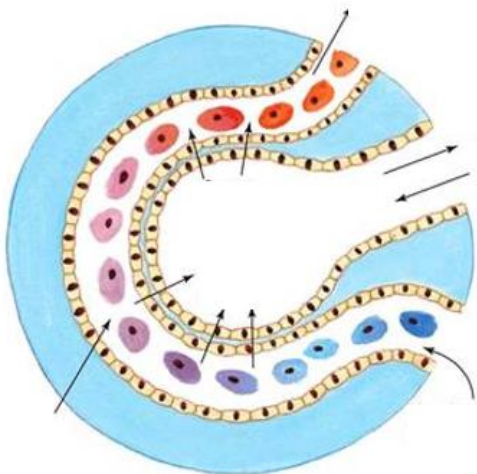
Ventilation and Gas Exchange: [https://www.youtube.com/watch?v=Y8R\\_fqJS9jA](https://www.youtube.com/watch?v=Y8R_fqJS9jA)

Q1. Label the diagram below of the lungs and ventilation system

## MEMORISE FOR RETRIEVAL TEST



Q2. Label the diagram of the alveolus and indicate the direction of the flow of gases.



Labels

Capillary	Oxygenated blood	Deoxygenated blood
Alveolus (air sac)	Oxygen	Carbon dioxide
Oxygen in	Carbon dioxide out	Short diffusion distance
From pulmonary artery	To pulmonary vein	One cell thick

Q3. What are the three main features of the gas exchange system?

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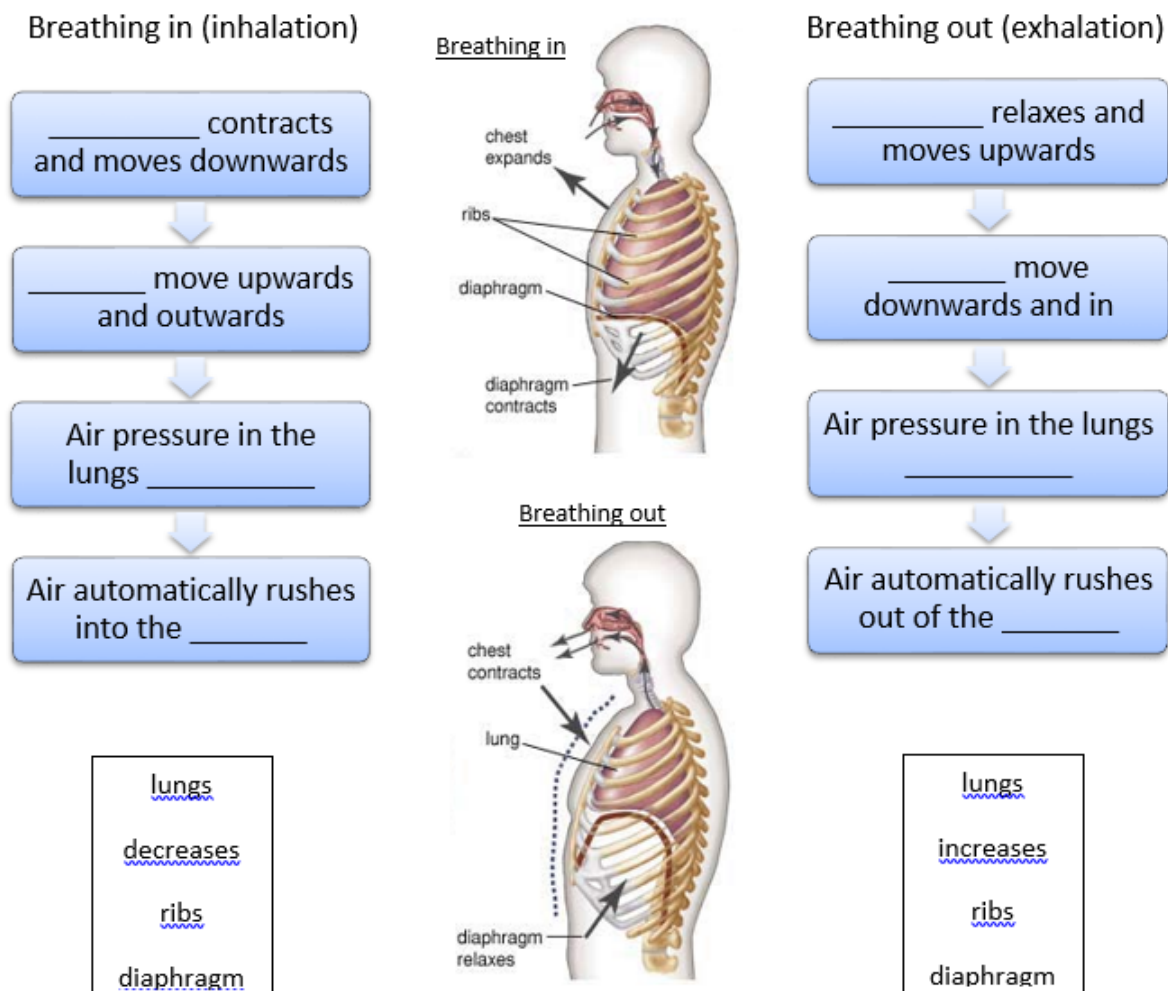
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Q4. Complete the flowchart showing inspiration and expiration

**MEMORISE FOR RETRIEVAL TEST**



## Apply

Describe and explain how the lungs of a mammal are adapted for rapid gas exchange.

(5)

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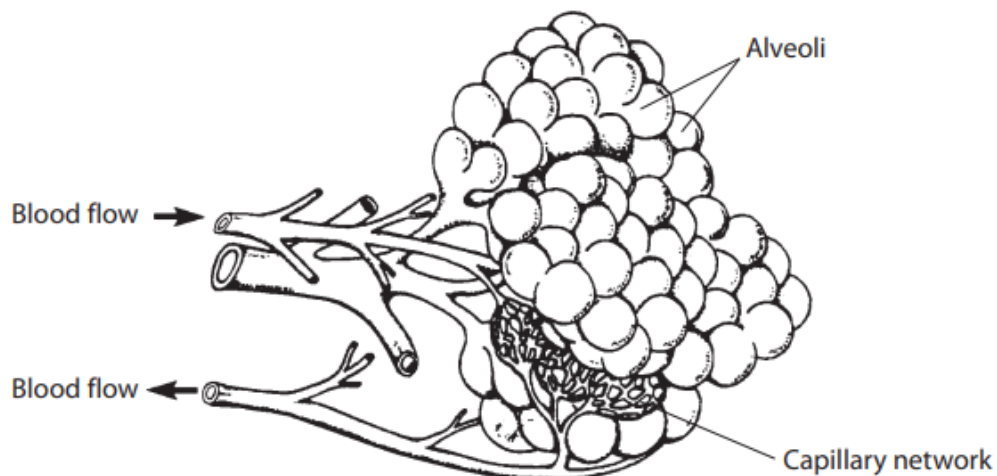
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The lungs in a mammal are adapted for efficient gas exchange.

- (a) The diagram below illustrates a small part of the lung responsible for gas exchange.



- (i) On the diagram, add a line labelled P to a branch of the pulmonary vein.

(1)

- (ii) Give **one** difference between the structure of a capillary and the structure of a vein.

(1)

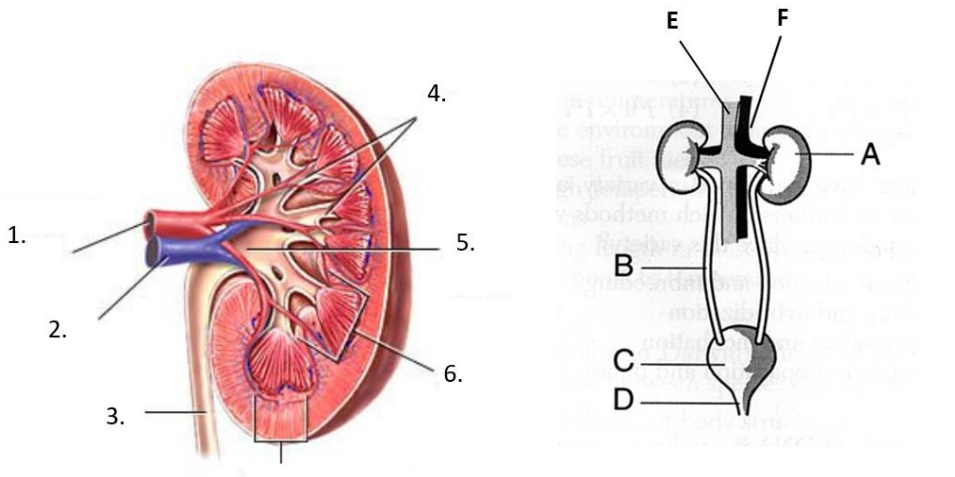
# Urinary System

Use the following links to support you with the questions

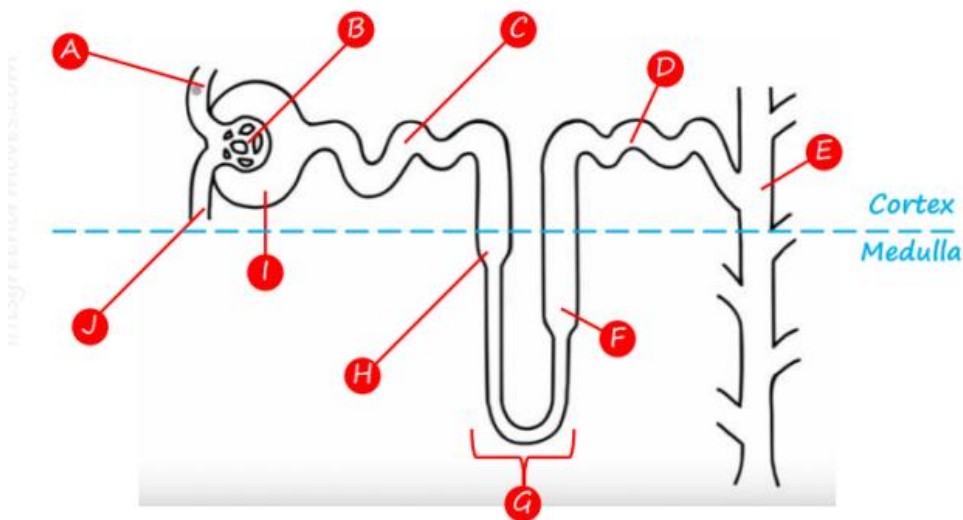
The Kidney: <https://www.savemyexams.com/igcse/biology/cie/23/revision-notes/13-excretion-in-humans/13-1-excretion-in-humans/13-1-2-the-kidney/>

The Nephron: <https://www.youtube.com/watch?v=tx9hYFeEd1E>

**Q1. Label the structures and diagrams below**



**Q2. Label the diagram of the nephron**





## Apply

(a) Explain what is meant by the term *homeostasis*.

[2]

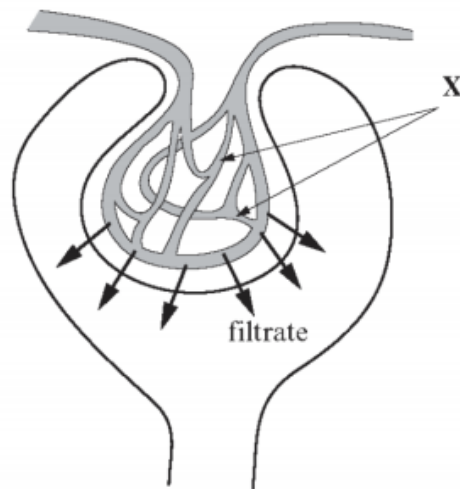
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The diagram shows part of a kidney tubule or nephron.



(a) (i) Name the network of capillaries labelled X. [1]

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(ii) Apart from water and glucose, name **two** substances which will be present in the filtrate. [1]

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(iii) Name the process that separates these molecules from the blood plasma. [1]

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# Cell Transport

Use the following links to support you with the questions

Cell Membrane Transport: <https://www.youtube.com/watch?v=J5pWH1r3pgU>

The Cell Membrane: <https://www.savemyexams.com/a-level/biology/cie/22/revision-notes/4-cell-membranes--transport/4-1-fluid-mosaic-membranes/4-1-3-the-cell-surface-membrane/>

## Surface Area to Volume Ratio

**Q1. Complete the table below**

Length of each side of cube (L) cm	Area of each face of cube (L <sup>2</sup> ) cm <sup>2</sup>	Surface area of the whole cube (L <sup>2</sup> ) × 6 cm <sup>2</sup>	Volume of the cube (L <sup>3</sup> ) cm <sup>3</sup>	Surface area to volume ratio $\frac{\text{surface area cm}^2}{\text{volume cm}^3}$
2				
4				

**Q2. Now work out what the surface area to volume ratio of a 10 cm × 10 cm cube is.**

Length of each side of cube (L) cm	Area of each face of cube (L <sup>2</sup> ) cm <sup>2</sup>	Surface area of the whole cube (L <sup>2</sup> ) × 6 cm <sup>2</sup>	Volume of the cube (L <sup>3</sup> ) cm <sup>3</sup>	Surface area to volume ratio $\frac{\text{surface area cm}^2}{\text{volume cm}^3}$
10				

**Q3. Complete the following sentences.**

The ..... the animal, the ..... the surface area : volume (SA/V) ratio.

Animals often have ..... surface areas in hot climates, and ..... surface areas in cold climates.

**Q4. Draw a diagram of the cell surface membrane, including phospholipid bilayer and channel proteins**

**Q5. Describe the different types of Passive Transport**

Diffusion	
Facilitated Diffusion	
Osmosis	

**Q6. Describe the process of Active Transport with the aid of a diagram**

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Q5. The fluid mosaic model has been developed from the knowledge of the structure and properties of cell membranes. It can explain how molecules can enter and leave a cell.

(5)

[illegible]

(b) Suggest **two** properties of molecules that enable them to enter a cell by diffusion.

(2)

1 .....

2 .....

(c) Facilitated diffusion and active transport are two ways in which molecules are transported across cell membranes.

Describe **one** similarity and **one** difference between facilitated diffusion and active transport.

(i) Similarity

(1)

.....  
.....  
.....  
.....

(ii) Difference

(1)

.....  
.....  
.....  
.....

# PHYSICS –

## NOTES (recall)

Watch the following [videos](#) and complete the notes on energy:

You don't need to include any derivations of formulae.



### Conservation of energy ([link](#))

State the law of conservation of energy

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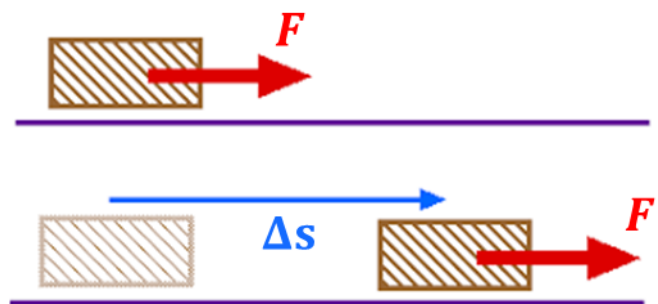
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### Mechanical work & energy ([link](#))

1. State the unit of energy \_\_\_\_\_

2. An object is pushed with a force,  $F$ , in the same direction as its displacement,  $\Delta s$ .

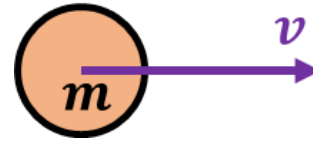
State the equation for the work done:



## Kinetic energy ([link](#))

1. What do we associate **kinetic** energy with? \_\_\_\_\_

2. An object of mass,  $m$ , is moving at a velocity,  $v$ .  
State the equation for the kinetic energy of the object:

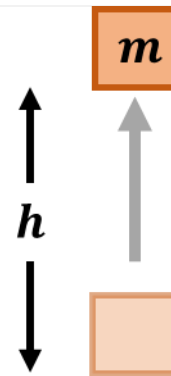


## Gravitational potential energy ([link](#))

1. How could you alter the gravitational potential energy of an object?

\_\_\_\_\_

2. An object of mass,  $m$ , is lifted a distance,  $h$ , in a uniform gravitational field,  $g$ .  
State the equation for the gravitational potential energy of the object:

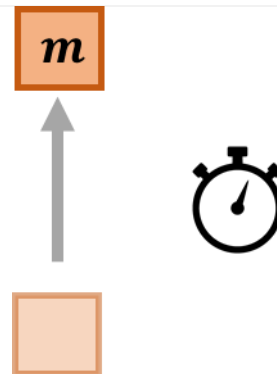


## Power ([link](#))

1. How could you be more powerful climbing a flight of stairs?

\_\_\_\_\_

2. Work,  $W$ , is done on an object moving shifting,  $E$ , between energy stores in a time,  $t$ .  
State the equation for the power transferred to the object:

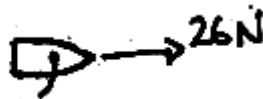


## Efficiency ([link](#))

1. State the equation for the efficiency of an object.

## APPLY - QUESTIONS (to be completed on separate paper)

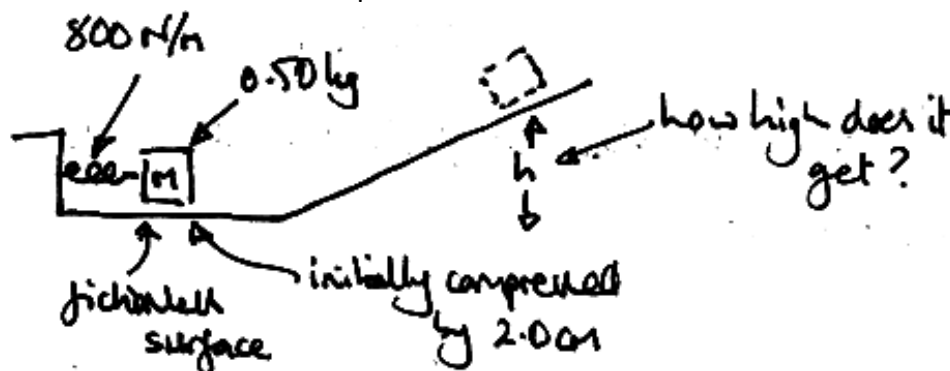
1. A toy boat moves 4.00 m in the direction of the applied force of 26 N on the sails. Calculate the work done on the boat.



2. Calculate the gravitational potential energy stored raising a 60 kg child to the top of a 2.0 m slide.



3. A spring with a constant of 25 N/m is extended by 50 cm. Calculate the stored elastic potential energy in the spring.
4. Calculate the kinetic energy of a 80 kg student sprinting at 8.0 m/s.
5. A 4.0 kg ball has 50 J of kinetic energy. How fast is it moving?
6. The child in question 2 'loses' 10 % of their gravitational potential energy store to frictional forces going down the slide. Calculate the speed of the child at the bottom of the slide.
7. A 10 g spring with a spring constant of 20 N/m is extended by 10 cm then released horizontally on a frictionless surface. Calculate the speed of the spring after release.
8. **(Challenge)** A 0.50 kg mass is launched from a frictionless surface by a spring with a spring constant of 800 N/m which is initially compressed by 2.0 cm. How high does the mass get up a slope as shown? Provide the vertical displacement.



9. A kettle transfers 1,500 J of energy, 1,200 J to a thermal energy store in the water and 300 J to the air as sound. How efficient is the kettle? **[2 marks]** [grade 3]
10. Playing on an Xbox 360 uses 102 J of electrical energy each second. 9 J of this energy is transferred to a thermal energy store. How efficient is the Xbox? **[3]** [grade 4]
11. A pendulum with a store of 100 J of gravitational potential energy transfers 99.5 J to a kinetic energy store on its first swing.
  - a. How much energy is being wasted due to friction by transfer to a thermal energy store? **[1]**
  - b. How efficient is the pendulum? **[2]** [grade 4]

PTO



## NOTES (preview)

Watch the following [videos](#) and complete the notes on materials:

You don't need to include any derivations of formulae.



### Hooke's law ([link](#))

1. Define tension

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2. What happens to objects under tension?

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3. Define compression

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4. What happens to objects under compression?

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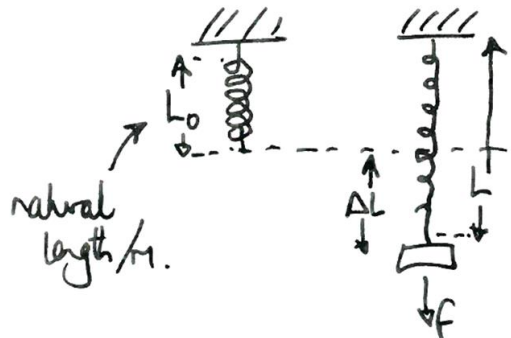
5. What is the additional length,  $\Delta L$ , ( $x$  in the video) called when a spring is stretched past its original (natural length?)

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6. What does the spring constant aka the stiffness,  $k$ , tell us?

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7. State the equation for Hooke's law:



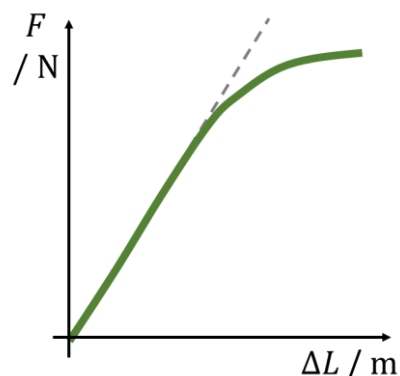
8. What does the gradient of the linear part of the curve represent?

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9. Name the point at which the graph is no longer linear.

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## Hooke's law energy ([link](#))

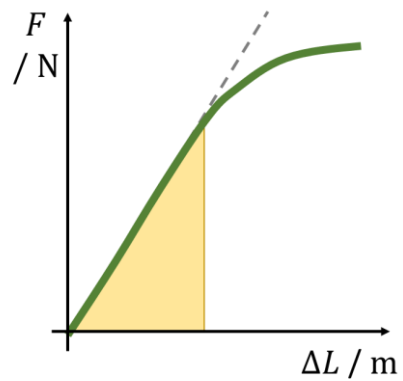
1. What does the area under the linear part of the curve represent?

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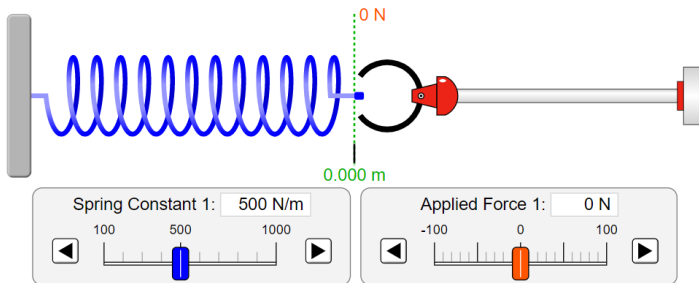
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2. State the equation(s) that there area represents.

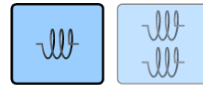


## Hooke's law simulation ([link](#))

Set up the simulation as shown in the images provided



- ☒ Applied Force →
- ☐ Spring Force →
- ☒ Displacement →
- ☒ Equilibrium Position
- ☒ Values



**SIMULATION**

1. With the spring constant set at 500 N/m apply a tension force of 50 N. Record your extension.

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2. Use Hooke's law to predict the extension when 100 N are applied.

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3. Apply a force of 100 N in the simulation. Record the extension.

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Stretch. Try adjusting the parameters of the simulation including viewing the elastic potential energy stored on the 3<sup>rd</sup> tab. Describe your findings:




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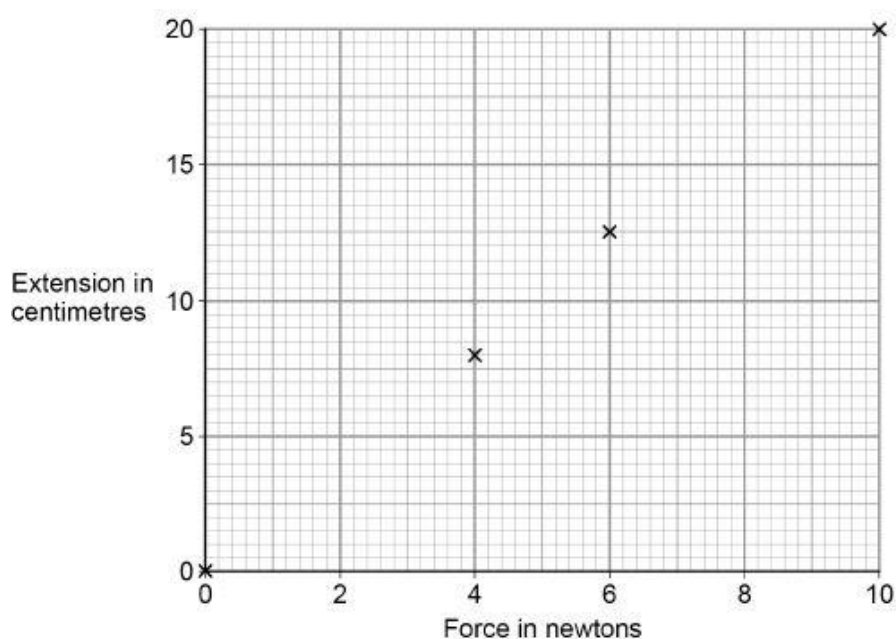
## APPLY - QUESTIONS (to be completed on separate paper)

**Q1.** A student carried out an investigation to determine the spring constant of a spring.

The table below gives the data obtained by the student.

Force in N	Extension in cm
0	0.0
2	3.5
4	8.0
6	12.5
8	16.0
10	20.0

The diagram below shows some of the data obtained by the student.



(c) Complete the diagram above by plotting the missing data from the table above.

Draw the line of best fit.

The table above is repeated here to help you answer this question.

(2)

(d) Write down the equation that links extension, force and spring constant.

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(1)

- (e) Calculate the spring constant of the spring that the student used.

Give your answer in newtons per metre.

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Spring constant = \_\_\_\_\_ N/m

**(4)**

- (f) Hooke's Law states that:  
'The extension of an elastic object is directly proportional to the force applied, provided the limit of proportionality is not exceeded.'

The student concluded that over the range of force used, the spring obeyed Hooke's Law.

Explain how the data supports the student's conclusion.

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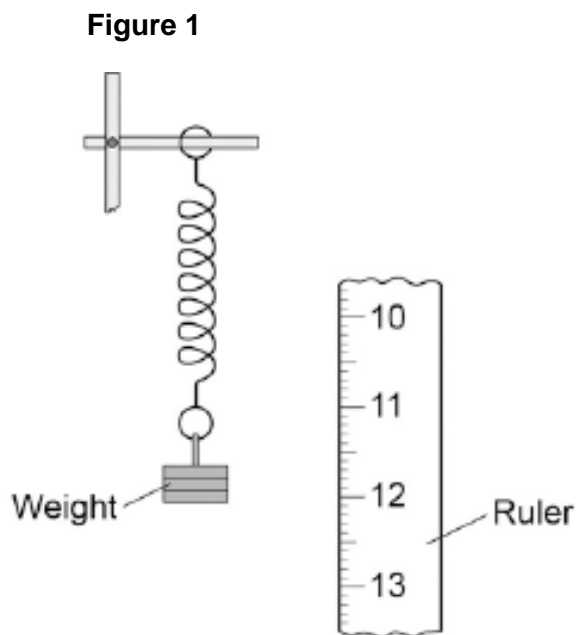
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**(2)**  
**(Total 9 marks)**

**Q2.** A student suspended a spring from a laboratory stand and then hung a weight from the spring.

**Figure 1** shows the spring before and after the weight is added.



- (a) Measure the extension of the spring shown in **Figure 1**.

Extension = \_\_\_\_\_ mm

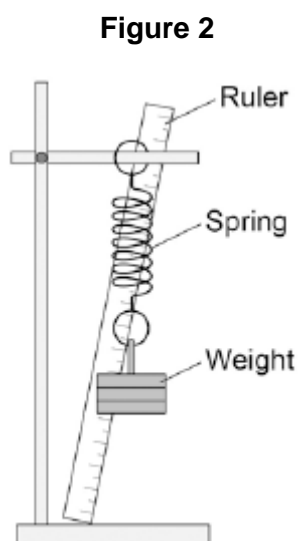
(1)

- (b) The student used the spring, a set of weights and a ruler to investigate how the extension of the spring depended on the weight hanging from the spring.

Before starting the investigation the student wrote the following prediction:

The extension of the spring will be directly proportional to the weight hanging from the spring.

**Figure 2** shows how the student arranged the apparatus.



Before taking any measurements, the student adjusted the ruler to make it vertical.

Explain why adjusting the ruler was important.

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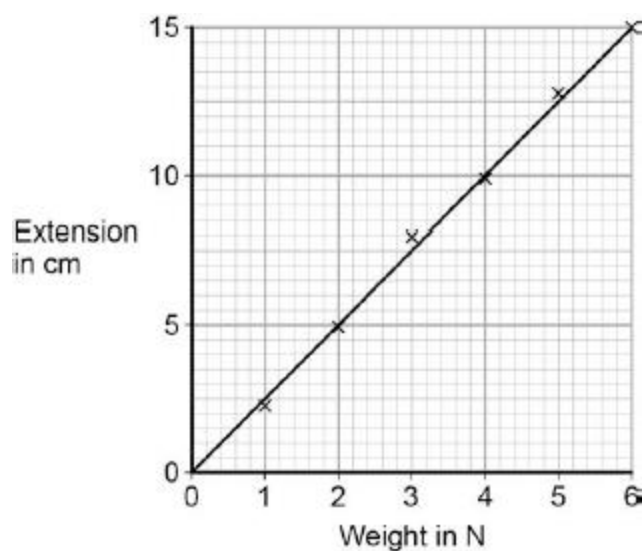
**(2)**

**PTO**

- (c) The student measured the extension of the spring using a range of weights.

The student's data is shown plotted as a graph in **Figure 3**.

**Figure 3**



What range of weight did the student use?

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(1)

- (d) Why does the data plotted in **Figure 3** support the student's prediction?

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(1)

- (e) Describe **one** technique that you could have used to improve the accuracy of the measurements taken by the student.

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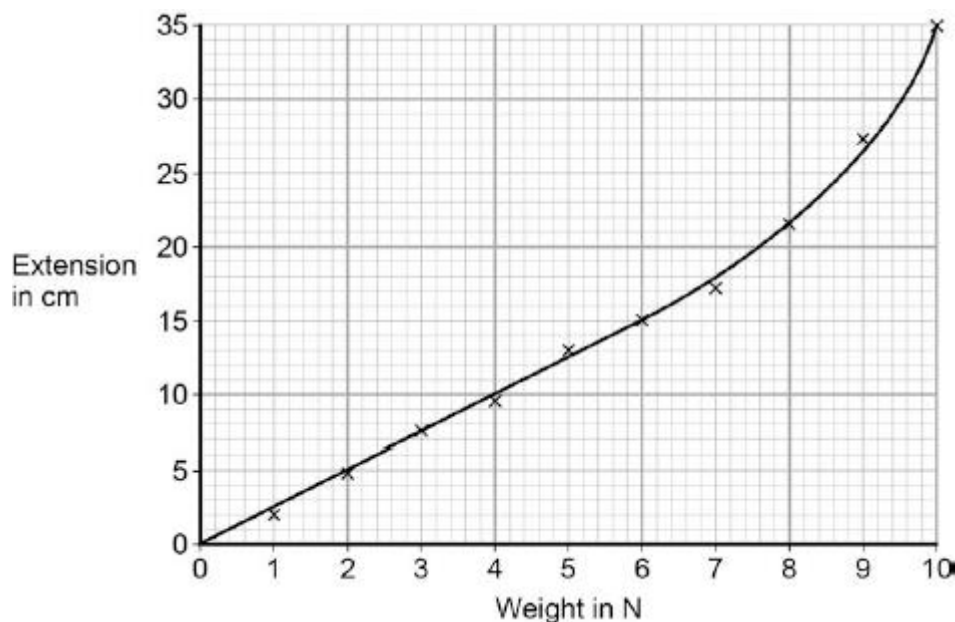
(2)

**PTO**

- (f) The student continued the investigation by increasing the range of weights added to the spring.

All of the data is shown plotted as a graph in **Figure 4**.

**Figure 4**



At the end of the investigation, all of the weights were removed from the spring.

What can you conclude from **Figure 4** about the deformation of the spring?

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Give the reason for your conclusion.

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(2)  
(Total 9 marks)



# CHEMISTRY

## Electrolysis

Give the definition of the following terms: (<https://www.bbc.co.uk/bitesize/guides/zcsyw6f/revision/1>)

i. Electrolysis

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ii. Electrolyte

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iii. Electrode

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iv. Reduction

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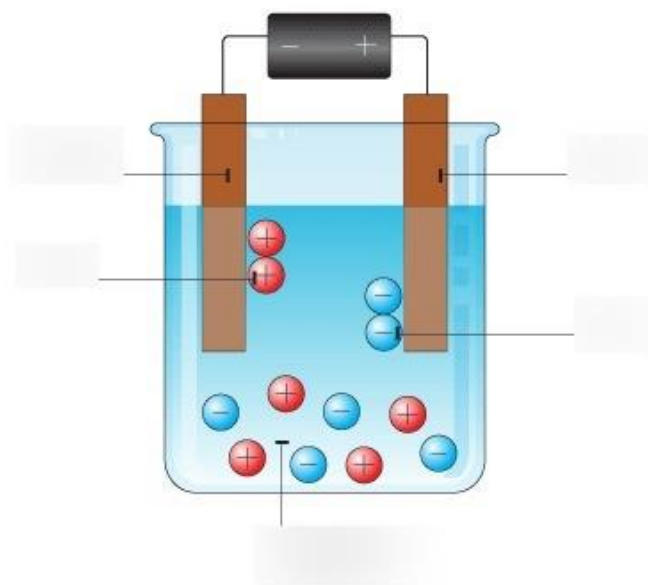
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v. Oxidation

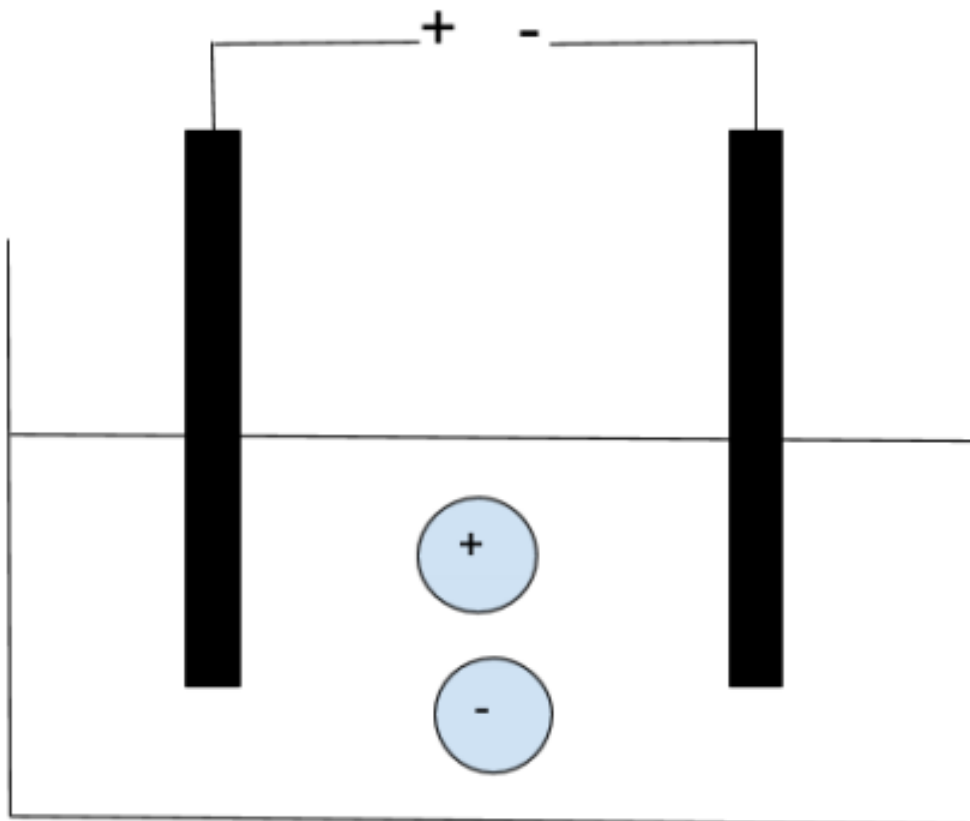
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Label the diagram (<https://www.sciencephoto.com/media/1156157/view/electrolysis-illustration>)



1. Complete the diagram with the following labels: anode, cathode, electrode, electrolyte, anion and cation



2. Describe an ionic bond.

.....

.....

3. Why can't a solid ionic substance conduct electricity?

.....

.....

4. Why can molten and solutions of ionic substances conduct electricity?

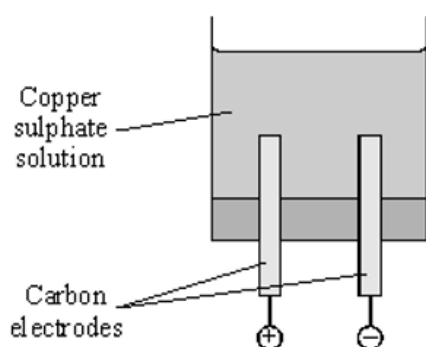
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Complete the table

Substance electrolysed	What is formed at the cathode	What is formed at the anode
molten lead iodide		
	Sodium	chlorine
Copper sulfate solution		
calcium bromide solution		
molten aluminium oxide		

**Q1.** An investigation into the *electrolyte* copper sulphate solution was carried out as shown.



(a) What does *electrolyte* mean?

.....

.....

.....

(2)

(b) These were the observations.

Negative electrode	solid formed
Positive electrode	gas given off

(i) Name the solid formed.

.....

(1)

(ii) Name the gas given off.

.....

(1)

(c) How could a sample of gas be collected at the positive electrode?

.....

.....

(2)

(d) Suggest why the blue colour of copper sulphate becomes paler during the investigation.

.....

.....

(2)

(Total 8 marks)

### **Ionic half equations**

A half-equation shows you what happens at one of the electrodes during electrolysis.

Electrons are shown as  $e^-$ .

A half-equation is balanced by adding, or taking away, a number of electrons equal to the total number of charges on the ions in the equation.

When positive metal ions (cations) arrive at the negative electrode (the cathode), they gain electrons to form neutral metal atoms. This is called reduction. For example:



## Q1.

A student makes a hypothesis:

'When different salt solutions are electrolysed with inert electrodes, the product at the negative electrode is always a metal'.

- (a) Describe how you would test this hypothesis in the laboratory.

You should:

- draw a labelled diagram of the apparatus
- give the independent variable
- describe what you would see at the negative electrode if the hypothesis is true.

Diagram

Independent variable \_\_\_\_\_

\_\_\_\_\_

Observation \_\_\_\_\_

\_\_\_\_\_

(5)

- (b) The student's hypothesis is only partially correct.

Explain why the product at the negative electrode is not always a metal.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

(2)

- (c) Predict the product at the positive electrode in the electrolysis of:

- sodium chloride solution
- copper sulfate solution.

Sodium chloride solution \_\_\_\_\_

Copper sulfate solution \_\_\_\_\_

(2)

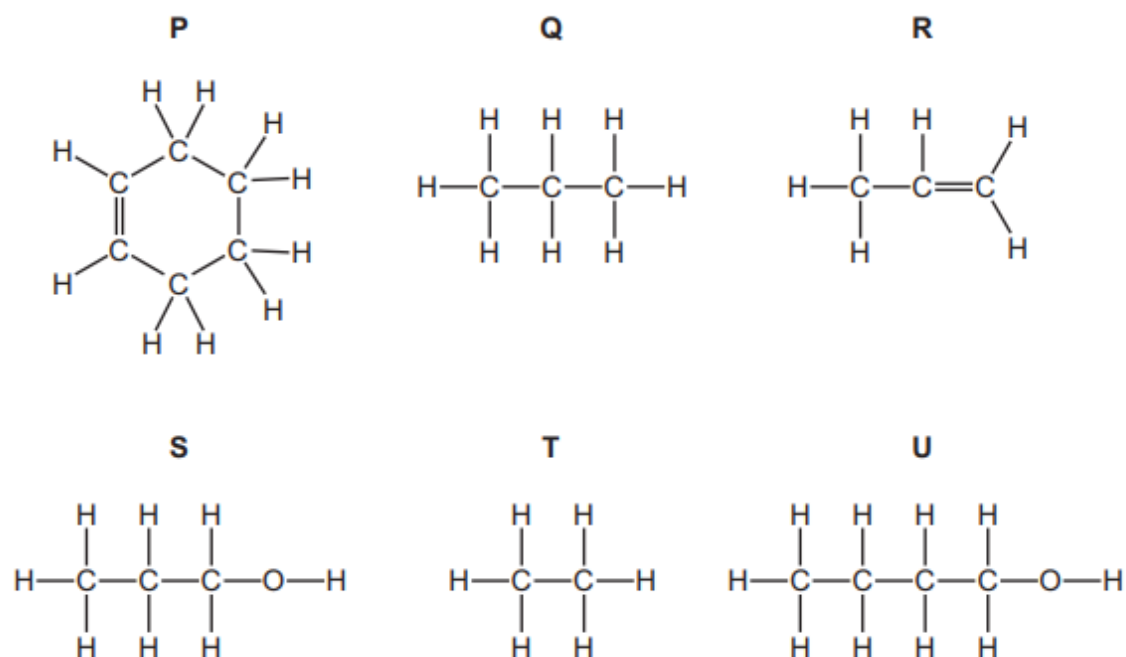
# Organic chemistry

Use the links to help you with this section:

1. <https://www.bbc.co.uk/bitesize/guides/zshvw6f/revision/2>
2. <https://www.bbc.co.uk/bitesize/guides/z3v4xfr/revision/1>
3. <https://www.bbc.co.uk/bitesize/guides/z3v4xfr/revision/3>

Name	Molecular Formula	Structure
Methane		
Ethane		
Propene		
Butane		
Pentene		
Hexane		

4 The structures of some organic compounds are shown.



- (a) (i) Which **two** of these compounds are alcohols?

Explain your answer.

.....  
.....

[2]

- (b) Methanol and ethanol are alcohols in the same homologous series.

Complete the following sentence about a homologous series using words from the list.

**alcohols**

**chemical**

**compounds**

**elements**

**functional**

**mixtures**

**physical**

A homologous series is a family of similar ..... with similar  
..... properties due to the same ..... group. [3]

# Nomenclature

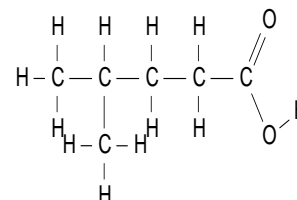
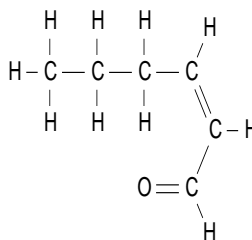
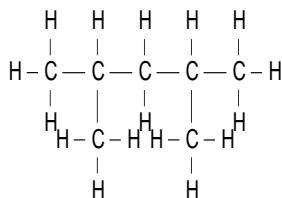
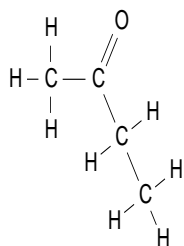
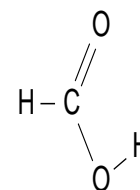
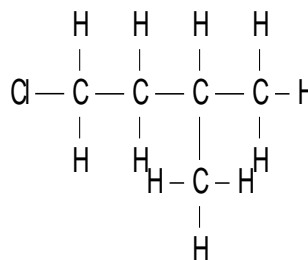
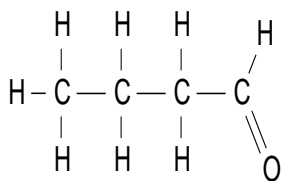
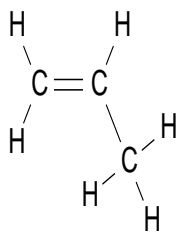
Use the following resources to help you with this section:

<https://chemrevise.org/wp-content/uploads/2021/02/3.1-revision-guide-introduction-organic-aqa.pdf> (page 3)

For each of the molecules below:

a) state the name of the molecule

b) state the functional group(s) present





# Part 2 – Highly Recommended

## BIOLOGY

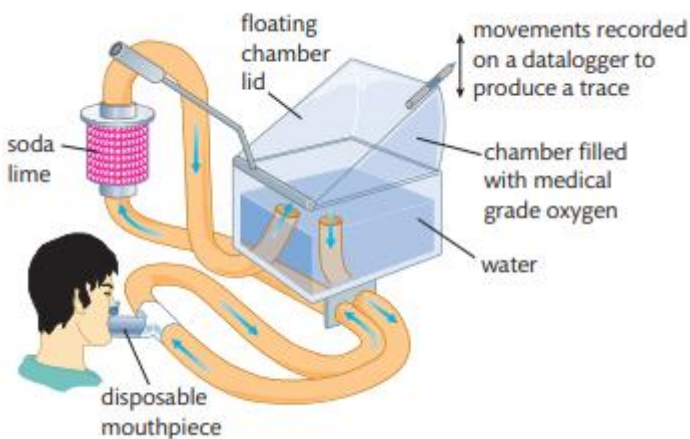
Use the following links to support you with the questions

Spirometry: <https://www.youtube.com/watch?v=GDRfYJlCxl>

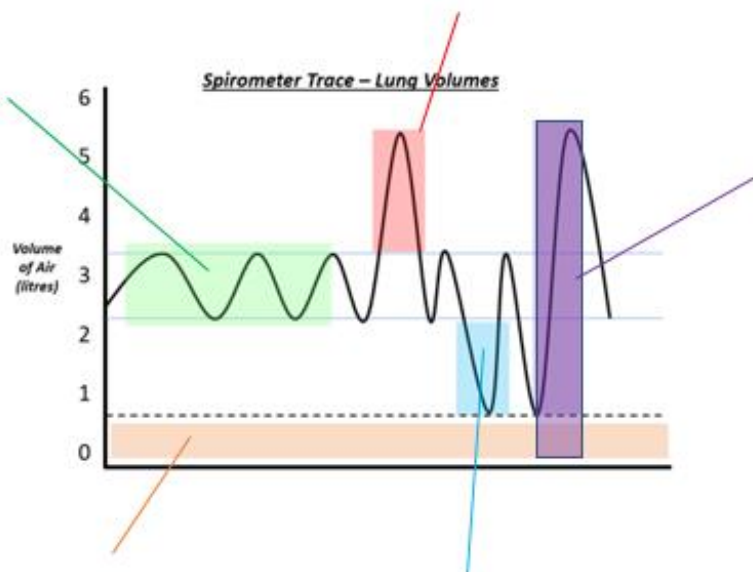
The Kidney: <https://www.savemyexams.com/igcse/biology/cie/23/revision-notes/13-excretion-in-humans/13-1-excretion-in-humans/13-1-2-the-kidney/>

The Heart: <https://www.savemyexams.com/qcse/biology/aqa/18/revision-notes/2-organisation/2-2-organisation-the-cardiovascular-respiratory-system/2-2-2-the-heart/>

**Q1. Describe the function of a Spirometer**



**Q2. Label the Spirometer Trace.**



**Q3. Complete the table below with the correct definitions**

<b>Tidal Volume</b>	
<b>Inspiratory Reserve Volume</b>	
<b>Expiratory Reserve Volume</b>	
<b>Total Lung Capacity</b>	
<b>Residual Volume</b>	

**Q3. Complete the paragraph below**

The kidney is a vital organ in the body and is responsible for excretion. It also plays an important role in homeostasis.

(a) Complete the passage, using the most suitable term in each case. The blood in the glomerulus has a high ..... pressure, which forces small molecules, such as glucose and ..... , out of the glomerulus and into the lumen of the Bowman's capsule. This process is known as ..... . In the proximal convoluted tubule, the glucose, most of the ..... and some of the salts are reabsorbed into blood ..... that surround the nephron at this point. [5]

(b) One aspect of the kidney's homeostatic role is the ability of anti-diuretic hormone (ADH) to increase the number of aquaporins in the plasma membranes of the cells lining the collecting duct. This increases the amount of water reabsorbed. ADH is released in response to a decrease in the water potential of the blood plasma.

(i) State precisely where the cells that detect a decrease in the water potential of the blood plasma are found.

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(ii) Name the cells that detect this decrease.

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**Q4. Calculate the cardiac output for an individual who has a heart rate of 70bpm and stroke volume of 132 ml**

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***Q5. Draw a flowchart below of the cardiac cycle***

## Apply

Q1. The atrio-ventricular node (AVN), bundle of His and Purkinje fibres are specialised cardiac muscle tissues which are involved in the control of heartbeat.

(a) State the function of the following structures in the functioning of the heart.

(i) atrio-ventricular node (AVN)

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(ii) bundle of His and Purkinje fibres

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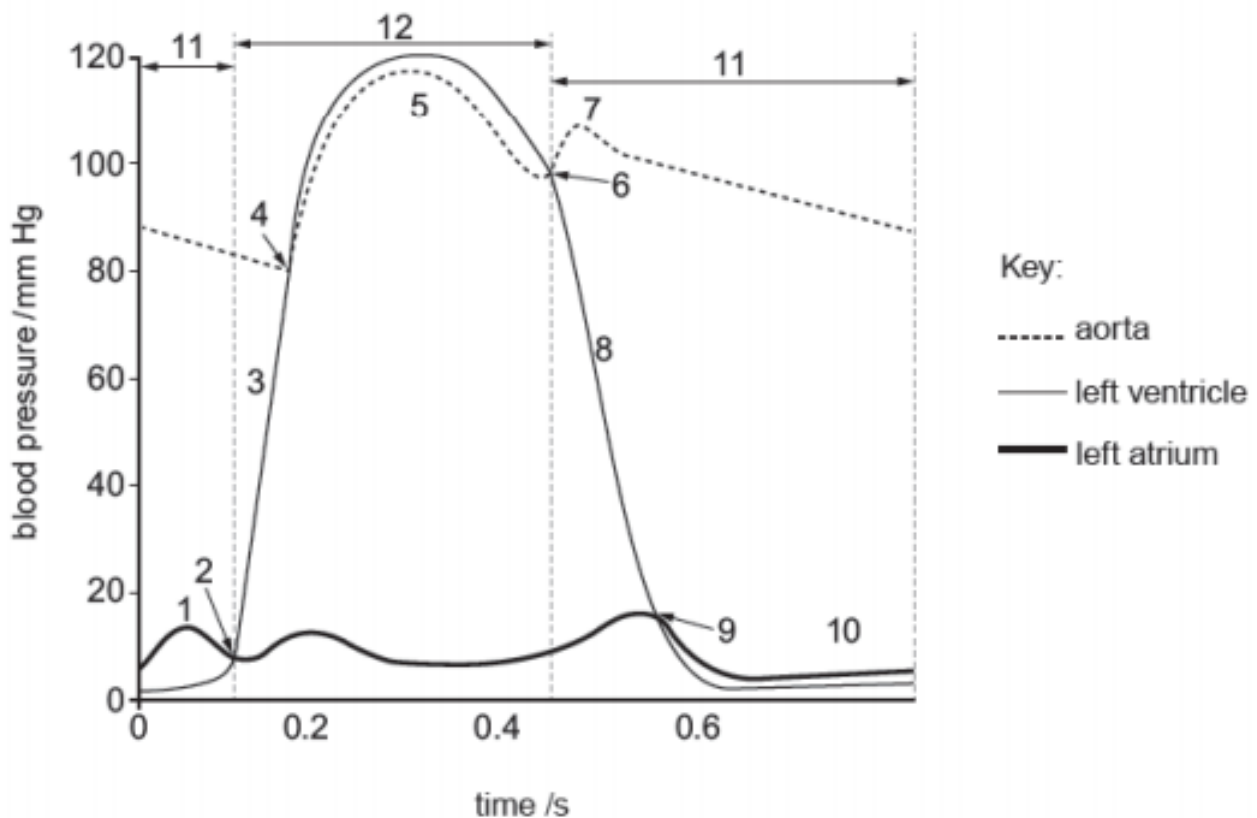
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The graph below shows the pressure changes in the left ventricle, left atrium and aorta during one cardiac cycle.



(b) The following statements list events or phases that occur during a cardiac cycle. State the numbers indicated on the graph above that correspond to each of the following statements. **PTO**

(i) ventricular diastole (ventricles are relaxing)

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(ii) recoil of aorta

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(iii) atrial systole

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(iv) closing of semi-lunar valves

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(v) opening of semi-lunar valves

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(vi) atrio-ventricular valves close

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(vii) ventricular systole (ventricles are contracting)

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(viii) passive filling of atrium by venous return

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Q2. Each kidney contains approximately one million nephrons. Each section of a kidney nephron is adapted to perform its function effectively. Describe the features of the glomerulus and Bowman's capsule that allow them to perform their function effectively

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Q3. (a) Describe how muscles in the thorax (chest) cause air to enter the lungs during breathing.

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(b) An athlete exercised at different rates on an exercise bicycle.

The table shows the effects of exercise rate on his breathing rate and tidal volume

Exercise rate / arbitrary units	Breathing rate / breaths minute <sup>-1</sup>	Tidal volume / dm <sup>3</sup>
0	14.0	0.74
30	15.1	1.43
60	15.3	1.86
90	14.5	2.34
120	15.1	2.76
150	14.8	3.25
180	21.5	3.21
210	25.7	3.23

(i) The athlete cycled at the particular exercise rate for 5 minutes before the relevant readings were taken. Explain why the readings were taken only after the athlete had been cycling for 5 minutes.

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(ii) Calculate the total volume of air taken into the lungs in one minute at an exercise rate of 120 arbitrary units.

(iii) Give two conclusions that can be drawn from the figures in the table.

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# CHEMISTRY

## Reactions of organic compounds: Free radical substitution and Electrophilic addition

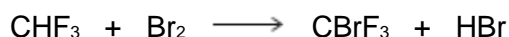
<https://chemrevise.org/wp-content/uploads/2023/10/3.2-revision-guide-alkanes-aqa.pdf>

(page 4-5)

### Q1.

There are many uses of halogenated organic compounds despite environmental concerns.

- (a) Bromotrifluoromethane is used in fire extinguishers in aircraft.  
Bromotrifluoromethane is formed when trifluoromethane reacts with bromine.



The reaction is a free-radical substitution reaction similar to the reaction of methane with chlorine.

- (i) Write an equation for each of the following steps in the mechanism for the reaction of  $\text{CHF}_3$  with  $\text{Br}_2$

Initiation step

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First propagation step

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Second propagation step

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A termination step

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(4)

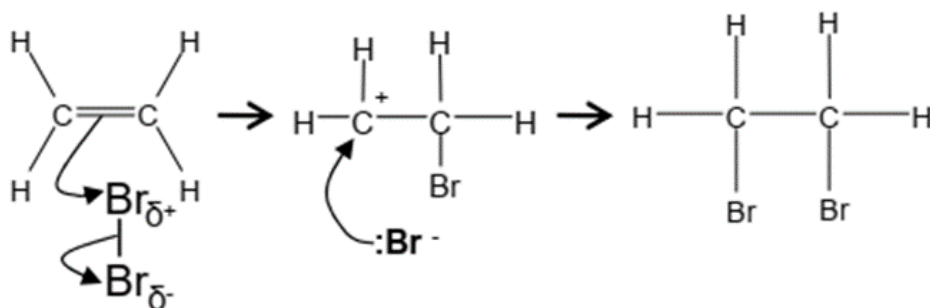
- (ii) State **one** condition necessary for the initiation of this reaction.

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(1)

## Example of the electrophilic addition mechanism

### Electrophilic addition of ethene to form 1,2-dibromoethane



Answer the following questions using the link:

<https://chemrevise.files.wordpress.com/2023/02/3.4-revision-guide-alkenes-aqa.pdf> (page 2-3)

1. Explain why alkenes tend to react with electrophiles.

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2. Draw out the mechanisms using curly arrows for the following reactions:

- i. ethene with bromine (Br<sub>2</sub>)

- ii. but-2-ene with hydrogen bromide (HBr)

- iii) ethene with concentrated sulphuric acid (H<sub>2</sub>SO<sub>4</sub>)



# PHYSICS –

## NOTES (preview)

Watch the following [videos](#) and complete the notes on materials:

You don't need to include any derivations of formulae.



### Stress and strain ([link](#))

1. State an equation for the stress,  $\sigma$ , and describe what it represents.

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2. State the unit of stress. \_\_\_\_\_

3. State an equation for the strain,  $\varepsilon$ , and describe what it represents.

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4. State the unit of strain. \_\_\_\_\_

5. Young's modulus is

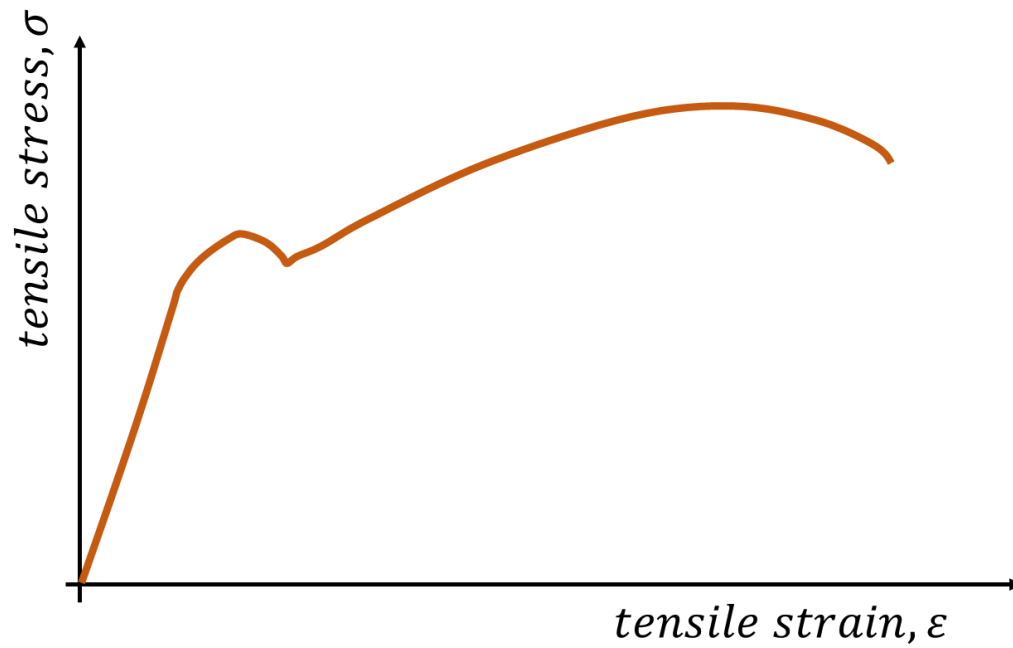
$$E = \frac{\sigma}{\varepsilon} = \frac{\text{stress}}{\text{strain}}$$

6. State the unit of Young's modulus. \_\_\_\_\_

7. Complete the sentence by selecting the appropriate words.

Youngs' elastic modulus,  $E$ , depends only upon the material / dimensions of the object under tension or compression. The larger the value of Young's modulus, the more / less stress is required to stain the material by a given amount. Materials with a high Young's modulus are stiff / flexible.

## Stress-strain graphs ([link](#))



1. Annotate points on the graph according to the video.
2. (stretch) Describe what happens at each point.

## Ductile and brittle materials ([link](#))

1. Describe what is meant by a brittle material.

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2. Describe what is meant by a ductile material.

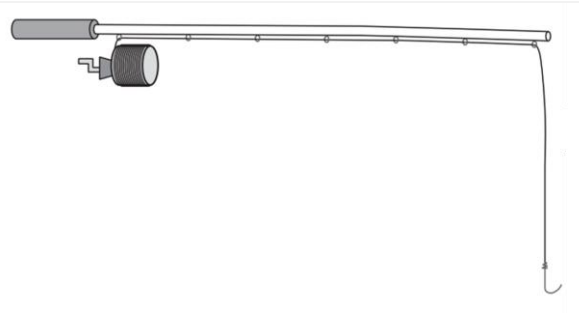
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## APPLY - QUESTIONS

1. Copper fishing lines are used for deep water fishing. A fish becomes caught on the hook and the copper line extends. Calculate the extension produced.

- Cross-sectional area of copper line =  $1.30 \times 10^{-7} \text{ m}^2$
- Load on line = 65 N
- Original length of line = 20.0 m
- Young's modulus of copper = 129 GPa



### Steps:

- Calculate the stress.
- Calculate the strain using Young's modulus equation.
- Calculate the extension using the strain equation.

2. A wire of a length 1.00 m and diameter of 0.40 mm is hung from a ceiling.

Find the extension caused in the wire, by attaching a weight of 100 N (2 s.f.), if the material of the wire has the Young's modulus of  $E = 200 \text{ GPa}$  (3 s.f.).

### Steps:

- Calculate the cross-sectional area in  $\text{m}^2$ .
- Calculate the stress in Pa.
- Calculate the strain using Young's modulus equation.
- Calculate the extension using the strain equation in m.